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Key indicators

Single-crystal X-ray study T = 133 K Mean σ (C–C) = 0.007 Å R factor = 0.038 wR factor = 0.039 Data-to-parameter ratio = 7.4

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

(1*R*,7*R*)-4-[(1*R*,2*S*,5*R*)-2-Isopropyl-5-methylcyclohexoxy]-9,9-dimethyl-2,2,6,6-tetraphenyl-3,5,8,10tetraoxa-4-phosphabicyclo[5.3.0]decane

The crystal structure of the title compound, $C_{41}H_{47}PO_5$, has been determined. The absolute configuration assignment was based on knowledge of the chirality of the starting material.

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Comment

Phosphorus-containing compounds are widely used as nonreactive ligands in metal catalysts. The title compound, (I), has been shown to induce enantioselectivity in Cu-mediated conjugate addition. This compound is also of interest as a ligand for Rh(I)-catalysed C—H bond activation. The bulky chiral phosphite is expected to induce enantioselectivity in the addition of C—H bonds across imine π -bonds.



The title compound contains a seven-membered heterocyclic backbone comprised of TADDOL (TADDOL = $\alpha, \alpha, \alpha', \alpha'$ -tetraphenyl-2,2'-dimethyl-1,3-dioxolane-4,5-dimethanol) and phosphorus (×2), the third substituent on phosphorus being a menthol bonded through oxygen. Bond lengths and angles are typical for organophosphites, and the structure of the TADDOL-phosphorus ring is similar to that of previously reported compounds (Keller *et al.*, 1998 and Sakaki *et al.*, 1993). In these previous examples, the third substituent on phosphorus has been relatively small and achiral (dimethylamine and phenyl, respectively). In the title compound, the larger chiral menthol provides an additional point of asymmetry, and is expected to increase the enantioselectivity of reactions carried out using metal complexes of this ligand.

Experimental

The title compound was prepared by the stepwise reaction of (+)-menthol and (-)-TADDOL with phosphorus trichloride, according to a literature procedure (Alexakis *et al.*, 2000). X-ray quality crystals were obtained by slowly cooling a concentrated solution of the title compound in an ether/pentane mixture to 238 K.

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Crystal data

C ₄₁ H ₄₇ O ₅ P $M_r = 650.79$ Orthorhombic, $P2_12_12_1$ a = 17.884 (1) Å b = 8.9233 (6) Å c = 22.395 (2) Å V = 3574.0 (4) Å ³ Z = 4 $D_x = 1.209$ Mg m ⁻³ Data collection	Mo $K\alpha$ radiation Cell parameters from 4403 reflections $\theta = 2.3-22.4^{\circ}$ $\mu = 0.12 \text{ mm}^{-1}$ T = 133.2 K Column, colorless $0.27 \times 0.20 \times 0.09 \text{ mm}$
Bruker SMART 1K CCD diffractometer ω scans Absorption correction: multi-scan (Blessing, 1995) $T_{min} = 0.967, T_{max} = 0.989$ 15797 measured reflections	5861 independent reflections 3158 reflections with $I > 3\sigma(I)$ $R_{int} = 0.069$ $\theta_{max} = 24.7^{\circ}$ $h = -21 \rightarrow 20$ $k = -9 \rightarrow 10$ $l = -22 \rightarrow 26$
Refinement Refinement on F R = 0.038 wR = 0.039 S = 0.96 3158 reflections	H-atom parameters constrained $w = 1/\sigma^2$ $(\Delta/\sigma)_{max} = 0.001$ $\Delta\rho_{max} = 0.21 \text{ e } \text{\AA}^{-3}$ $\Delta\rho_{min} = -0.32 \text{ e } \text{\AA}^{-3}$

Table 1

424 parameters

Selected geometric parameters (Å, °).

P1-01	1.610 (3)	P1-O3	1.656 (3)
P1-O2	1.629 (3)		
O1-P1-O2	93.8 (1)	O2-P1-O3	99.8 (1)
O1-P1-O3	95.9(1)		

When the absolute configuration analysis was performed in teXsan, very little change was observed in calculated *versus* observed intensity differences between Friedel pairs upon inversion. The correct enantiomer shows 1225 pairs of reflections with agreement between calculated and observed intensities, and 1184 in disagreement. The known chirality of the starting materials were therefore used nto assign the enantiomer.

Data collection: *SMART* (Bruker, 1995–1999); cell refinement: *SAINT* (Bruker, 2002); data reduction: *SAINT*; program(s) used to solve structure: *SIR*92 (Altomare *et al.*, 1993); program(s) used to refine structure: *teXsan* (Molecular Structure Corporation, 1985, 1992); molecular graphics: *ORTEP*-3 (Farrugia, 1997); software used to prepare material for publication: *teXsan*.

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Figure 1

The title compound, with displacement ellipsoids drawn at the 50% probability level. H atoms are omitted for clarity.

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